

## Palladium Catalysis

International Edition: DOI: 10.1002/anie.201911961 German Edition: DOI: 10.1002/ange.201911961

# Asymmetric Wacker-Type Oxyallenylation and Azaallenylation of Cyclic Alkenes

Shenghan Teng, Zhiwei Jiao, Yonggui Robin Chi, and Jianrong Steve Zhou\*

**Abstract:** Palladium-catalyzed three-component carboetherification of cyclic alkenes proceeded to give trans adducts exclusively with excellent enantioselectivity through a Wackertype pathway. The reaction is also applicable to other oxygen nucleophiles, such as water, phenols, and carboxylic acids, as well as some electron-poor aryl amines.

Catalytic carboetherification and carboamination of alkenes allows the construction of a new C-C bond and a new C-O/N bond in a single reaction.<sup>[1]</sup> The three-component reaction is particularly attractive in terms of convergence and the possibility to establish two new stereocenters. Many existing three-component examples involve carbon-radical addition to alkenes as the key step.<sup>[2]</sup> The development of such an asymmetric radical reaction still remains a challenge. Until now, most examples of asymmetric carboetherification and carboamination have been limited to two-component reactions initiated by intramolecular heteropalladation<sup>[3]</sup> or heterocupration<sup>[4]</sup> (Figure 1a). In a similar strategy, Mazet and co-workers strategically placed hydroxy and sulfonamide groups at ortho positions of aryl palladium species, which set the stage for intramolecular syn attack at a bound cycloalkene, 2,3-dihydrofuran.<sup>[5]</sup> In an oxy-Heck-Matsuda reaction developed by Correia et al.,<sup>[6]</sup> aryl insertion produced an  $\eta^3$ benzyl complex, which, in turn, was intercepted by a pendant phenol to produce dihydrobenzofurans. The tethering strategy was also utilized by Wolfe and co-workers for asymmetric anti attack of external amines on an allylphenyl Pd species to form 2-amidoindanes.<sup>[7]</sup> Furthermore, Liu and co-workers recently disclosed a truly three-component copper-catalyzed asymmetric arylsulfonamidation of styrene, which proceeded by amino-radical addition to styrene and subsequent formation of a new stereocenter at the benzylic position.<sup>[8]</sup>

Herein, we report the first three-component example of asymmetric oxyallenylation,<sup>[9]</sup> using propargylic acetates and oxygen nucleophiles, such as alcohols, phenol, carboxylic

[\*] Prof. Dr. J. S. Zhou

State Key Laboratory of Chemical Oncogenomics, Key Laboratory of Chemical Genomics, School of Chemical Biology and Biotechnology Peking University Shenzhen Graduate School
2199 Lishui Road, Room F-312, Nanshan District, Shenzhen 518055 (China)
E-mail: jrzhou@pku.edu.cn
S. Teng, Dr. Z. Jiao, Prof. Dr. Y. R. Chi Division of Chemistry and Biological Chemistry, School of Physical

- and Mathematical Sciences, Nanyang Technological University, 21 Nanyang Link, Singapore 637371 (Singapore)
- Supporting information and the ORCID identification number(s) for
   the author(s) of this article can be found under: https://doi.org/10.1002/anie.201911961.



*Figure 1.* a) Use of tethering for asymmetric induction in the oxy- or azapalladation of alkenes. b) Plausible pathways of oxy- and azaalleny-lation and side reactions.

acids, and water (Figure 1b). The reactivity can also be extended to electron-deficient aryl amines. The reaction led to exclusive *trans* heteroallenylation in excellent enantiose-lectivity. Allenes are important intermediates in organic synthesis, especially so with the advent of many metal-catalyzed stereoselective and/or regioselective reactions to prepare substituted allenes.<sup>[10]</sup>

To produce the Wacker-type adducts, several side reactions must be prevented (Figure 1 b): 1) The  $\sigma$ -allenyl complex of palladium must undergo fast Wacker-type *anti* heteropalladation of alkenes instead of *syn* insertion<sup>[11]</sup> or Heck *syn* carbopalladation, the latter leading to fused cyclobutenes, which we reported previously.<sup>[12]</sup> 2) The allenyl complex may exist in equilibrium with the  $\pi$ -propargyl complex, which can eliminate a  $\beta$ -hydrogen atom to give an enyne. 3) The  $\pi$ -propargyl complex is electrophilic and can undergo propargylic substitution if nucleophilic amines are used.

Tethering of alkenes to the inserting heteroatoms or metal-bound aryl rings is a commonly employed strategy to facilitate asymmetric insertion (see Figure 1 a).<sup>[13]</sup> For simple alkenes that bind to catalysts through a single point of

contact, stereocontrol by catalysts still remains a challenge. Therefore, there has been no asymmetric example of the archetypical Wacker *anti* attack on simple monoolefins.

In our recent study of a model Heck annulation of 3phenylpropargyl acetate 2a and 2,3-dihydrofuran (Scheme 1), a new product of alkoxyallenylation 3a was obtained in 68 % yield through the reaction of the cosolvent, isopropanol in the presence of Pd(dba)<sub>2</sub> and dppp. A small amount of the enyne by-product was also detected.



**Scheme 1.** Effect of bisphosphines on a model alkoxyallenylation of 2,3-dihydrofuran (yields and *ee* values of **3 a** are indicated). dba=dibenzylideneacetone, Xyl = m-xylyl.

Encouraged by this finding, we then screened chiral bisphosphines to develop an enantioselective variant. In most cases,  $\beta$ -elimination to the envne was the main side reaction, whereas Heck annulation was rarely seen, except with the ligand Norphos L4 (23% yield). For example, biaryl diphosphines, such as BINAP, Segphos and Difluorphos gave the desired product 3a in less than 5% yield. Notably, the trifluoromethylated Josiphos ligand L7 provided 3a in 40% yield with 49% ee (and 28% elimination). Josiphos ligand L8 bearing electron-withdrawing 2-furyl groups afforded 3a in 90% yield with 24% ee, and notably without any elimination. Gratifyingly, a palladium catalyst with furyl-MeOBIPHEP L10 afforded 3a in 85% yield with 84% ee at 60°C. Moreover, 90% ee and 80% yield were observed at 45°C in the presence of 0.1 mol % Pd catalyst. Importantly, no enyne was detected. In comparison, catalysts formed by other MeOBI-PHEP ligands L11-L14 containing parent P-phenyl and electron-donating P-aryl substituents had almost no activity, and elimination was the main side reaction. As compared to triphenylphosphine, 2-furylphosphines are much weaker σdonors to transition metals and better  $\pi$ -acceptors.<sup>[14]</sup> They form an electronically deficient allenvl palladium center, which activates the bound alkene for Wacker attack by nucleophiles. This trend is consistent with the higher Wacker reactivity observed with electron-deficient Josiphos **L7** and **L8** as compared to **L6**. Notably, both MeOBIPHEP and Josiphos form quite large bite angles of 94° and 97°, respectively, at palladium(II) centers,<sup>[15]</sup> which may also help to prevent elimination. The corresponding monoxide of bisphosphine **L10** was also prepared and tested, but it only led to **3a** in low yield with less than 10% *ee*.<sup>[16]</sup>

The palladium catalyst (0.2 mol%) of L10 was then applied to reactions of 2,3-dihydrofuran with other propargylic acetates bearing both electron-rich and electron-deficient aryl rings in the 3-position (Scheme 2b). Esters, ketones, acetals, aryl bromides, fluorides, and chlorides were well tolerated. Furthermore, the reaction tolerated quinoline, thiophene, and indole (products 3p-r). When 3-aryl propargylic acetates containing 1,1-diethyl and cycloalkyl substituents were used, 1-2 mol% of the Pd catalyst was enough to give full conversion (products 3t-w). Unfortunately, 3alkyl propargylic acetates only afforded the adducts in low yields. Other alcohols, including benzyl alcohol, cyclohexanol, tert-butyl alcohol, and ethylene glycol also reacted smoothly (Scheme 2c, products 4a-f). The reaction of racemic propargylic acetate 2b led to a mixture of two allenyl isomers 4g, both with 88% ee (Scheme 2d). However, primary and secondary propargylic acetates did not react at all. The racemic 2,3-dihydrofuran 1b substituted with a p-methoxvphenyl ring afforded the single diastereomer 4h with 80% ee. The observed enantiofacial selectivity translates to an s value of 9 (Scheme 2e). Unfortunately, in the reaction of 2,3-dihydropyran, only simple addition of isopropanol to the alkene group was detected.

*N-tert*-butoxycarbonyl-2,3-dihydropyrrole (1c) was another reactive alkene in this transformation (Scheme 3), whereby different aryl rings at the 3-position of the propargylic acetate were tolerated (products **5b–f**), and both primary and secondary alcohols coupled in good yields (products **5i–m**). Notably, simple addition of alcohols to the double bond was also detected, accounting for the rest of the material. Reactions of cyclopentene afforded only a small amount of the desired products, however.

When **1a** was treated with acetic acid and phenol in acetonitrile, the *trans* adducts **6a** and **6b** were also generated exclusively (Scheme 4 a,b). The reaction with water generated **6c** as both *trans* and *cis* isomers in a ratio of 4:1 (Scheme 4 c), whereas a similar reaction of *N*-Boc-substituted dihydropyrrole **1c** afforded exclusively *trans* isomer **6d** (Scheme 4 d). The minor *cis*-lactol **6c** was probably formed through reversible ring opening of the initially formed *trans* lactol. Notably, adduct **6a** was also detected in small amounts in reaction mixtures of alkoxyallenylation in Scheme 2, but purified **6a** remained unreactive when subjected to the conditions of catalytic alkoxyallenylation. Thus, **6a** was chemically incompetent as an intermediate for the alkoxyallenylation process.

We also attempted to replace alcohols with amine nucleophiles in acetonitrile. When aniline and indoline were used, propargylic amination of 2a was the main side reaction (see Figure 1b). Moreover, aliphatic amines, such as diethyl-



## **Communications**







amine and pyrroline, led to elimination of 2a to give the enyne. However, fluorinated aryl amines and pyridyl amines of attenuated nucleophilicity reacted smoothly to provide azaallenylation adducts (Scheme 5 a,b). The initially formed *trans N,O*-acetals **6e,f** probably underwent reversible ring opening to generate the minor *cis* isomers. NaOTf accelerated the addition of pyridyl amines to give products **6g,h** 



Scheme 3. Alkoxyallenylation of N-Boc-2,3-dihydropyrrole.



**Scheme 4.** Wacker-type oxyallenylation using acetic acid, phenol, and water.



 $\label{eq:scheme 5. Wacker-type azaallenylation triggered by the attack of an aryl amine. Tf=trifluoromethanesulfonyl.$ 

(Scheme 5b); without this additive, no conversion was detected.

To demonstrate the synthetic utility of the transformation, we subjected adduct **3a** to Friedel–Crafts allylation and heteroarylation with an allylsilane<sup>[17]</sup> and furan (Scheme 6a). Acetal **3a** was also reduced to **7b** by triethylsilane, and then underwent facile iodocyclization to **7c** upon treatment with



**Scheme 6.** Transformations of alkoxyallenylation adducts. Ts = toluene-sulfonyl.

*N*-iodosuccinimide (NIS).<sup>[18]</sup> Acidic hydrolysis of **3a** and PCC oxidation furnished butyrolactone **7e** (Scheme 6b). In all cases, no erosion of enantiopurity was observed. Next, we attempted arylation of the oxonium species derived from **3a** under Friedel–Crafts conditions. Surprisingly, scandium triflate efficiently catalyzed the addition of 1,3,5-trimethoxybenzene to **3a** but giving a highly strained, fused cyclobutane **7f** through an unprecedented sequence of cationic reactions (Scheme 6b). A bromophenylated acetal **31** was also readily converted into **7h**, whose configuration was established by X-ray diffraction analysis (Scheme 6b).<sup>[19]</sup> However, the analogous Friedel–Crafts reaction of **5a** resulted in simple arylation (Scheme 6c). A thermal Diels–Alder reaction of **3s** with an *N*-aryl maleimide produced a mixture of two isomers of **7j** (Scheme 6d).

Adduct **5a** was also subjected to silane reduction and selective ozonolysis to give ketone **71** (Scheme 7).<sup>[20]</sup> Subsequent palladium-catalyzed reduction with polymethylhydro-



**Scheme 7.** Synthesis of chiral 3-benzoyl- and 3-benzylpyrrolidine. cod = 1,5-cyclooctadiene.

Angew. Chem. Int. Ed. 2020, 59, 2246-2250

siloxane (PMHS) afforded *N*-(*tert*-butoxycarbonyl)benzylpyrrolidine **7m**.<sup>[21]</sup> Chiral 3-benzylpyrrolidines are one type of heterocyclic amines used in drug discovery,<sup>[22]</sup> but few efficient stereoselective methods for their synthesis exist to date.<sup>[23]</sup>

In summary, we have reported an asymmetric Wackertype coupling of propargylic acetates, cyclic alkenes, and external nucleophiles, such as alcohols, phenols, carboxylic acids, water, and electron-deficient aryl amines. For the first time, asymmetric Wacker-type attack of truly external nucleophiles on monoolefins proceeded without any aid of directing groups. Notably, the use of electron-deficient furyl-MeOBIPHEP proved crucial to both the reactivity and stereoselectivity of the asymmetric Wacker-type process.

#### Acknowledgements

We acknowledge financial support from Peking University Shenzhen Graduate School, Shenzhen Bay Laboratory (21230011-Scripps), Nanyang Technological University, the GSK-EDB Trust Fund (2017 GSK-EDB Green and Sustainable Manufacturing Award), and A\*STAR Science and Engineering Research Council (AME IRG A1783c0010). We thank Dr. Li Yongxin for X-ray diffraction analysis of several compounds.

### Conflict of interest

The authors declare no conflict of interest.

**Keywords:** allenes  $\cdot$  asymmetric catalysis  $\cdot$  carboetherification  $\cdot$  palladium catalysis  $\cdot$  Wacker reaction

How to cite: Angew. Chem. Int. Ed. 2020, 59, 2246–2250 Angew. Chem. 2020, 132, 2266–2270

- For reviews, see: a) R. I. McDonald, G. Liu, S. S. Stahl, *Chem. Rev.* 2011, *111*, 2981; b) Z. J. Garlets, D. R. White, J. P. Wolfe, *Asian J. Org. Chem.* 2017, *6*, 636; c) J.-S. Zhang, L. Liu, T. Chen, L.-B. Han, *Chem. Asian J.* 2018, *13*, 2277.
- [2] For examples, see: a) S. Kirchberg, R. Fröhlich, A. Studer, Angew. Chem. Int. Ed. 2010, 49, 6877; Angew. Chem. 2010, 122, 7029; b) A. D. Melhado, W. E. Brenzovich, A. D. Lackner, F. D. Toste, J. Am. Chem. Soc. 2010, 132, 8885; c) G. Fumagalli, S. Boyd, M. F. Greaney, Org. Lett. 2013, 15, 4398; d) Y. Xie, J. Hu, P. Xie, B. Qian, H. Huang, J. Am. Chem. Soc. 2013, 135, 18327; e) M. N. Hopkinson, B. Sahoo, F. Glorius, Adv. Synth. Catal. 2014, 356, 2794; f) T. Itoh, Y. Shimizu, M. Kanai, Org. Lett. 2014, 16, 2736; g) H. Yi, X. Zhang, C. Qin, Z. Liao, J. Liu, A. Lei, Adv. Synth. Catal. 2014, 356, 2873; h) C. Chatalova-Sazepin, Q. Wang, G. M. Sammis, J. Zhu, Angew. Chem. Int. Ed. 2015, 54, 5443; Angew. Chem. 2015, 127, 5533; i) Z. Liao, H. Yi, Z. Li, C. Fan, X. Zhang, J. Liu, Z. Deng, A. Lei, Chem. Asian J. 2015, 10, 96; j) A. Tlahuext-Aca, R. A. Garza-Sanchez, F. Glorius, Angew. Chem. Int. Ed. 2017, 56, 3708; Angew. Chem. 2017, 129, 3762; k) S. Zheng, Á. Gutiérrez-Bonet, G. A. Molander, Chem 2019, 5, 339.
- [3] For examples, see: a) L. F. Tietze, K. M. Sommer, J. Zinngrebe, F. Stecker, Angew. Chem. Int. Ed. 2005, 44, 257; Angew. Chem. 2005, 117, 262; b) Y. Kawamura, Y. Kawano, T. Matsuda, Y. Ishitobi, T. Hosokawa, J. Org. Chem. 2009, 74, 3048; c) D. N.

© 2019 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim

Mai, J. P. Wolfe, J. Am. Chem. Soc. 2010, 132, 12157; d) B. A.
Hopkins, J. P. Wolfe, Angew. Chem. Int. Ed. 2012, 51, 9886;
Angew. Chem. 2012, 124, 10024; e) B. A. Hopkins, Z. J. Garlets,
J. P. Wolfe, Angew. Chem. Int. Ed. 2015, 54, 13390; Angew.
Chem. 2015, 127, 13588; f) N. Hu, K. Li, Z. Wang, W. Tang,
Angew. Chem. Int. Ed. 2016, 55, 5044; Angew. Chem. 2016, 128, 5128.

- [4] a) W. Zeng, S. R. Chemler, J. Am. Chem. Soc. 2007, 129, 12948;
  b) L. Miao, I. Haque, M. R. Manzoni, W. S. Tham, S. R. Chemler, Org. Lett. 2010, 12, 4739; c) Y. Miller, L. Miao, A. S. Hosseini, S. R. Chemler, J. Am. Chem. Soc. 2012, 134, 12149;
  d) M. T. Bovino, T. W. Liwosz, N. E. Kendel, Y. Miller, N. Tyminska, E. Zurek, S. R. Chemler, Angew. Chem. Int. Ed. 2014, 53, 6383; Angew. Chem. 2014, 126, 6501; e) S. D. Karyakarte, C. Um, I. A. Berhane, S. R. Chemler, Angew. Chem. Int. Ed. 2018, 57, 12921; Angew. Chem. 2018, 130, 13103.
- [5] a) V. Bizet, G. M. Borrajo-Calleja, C. Besnard, C. Mazet, ACS Catal. 2016, 6, 7183; b) G. M. Borrajo-Calleja, V. Bizet, C. Mazet, J. Am. Chem. Soc. 2016, 138, 4014.
- [6] C. R. Correia, A. Ribeiro, E. Polo, N. Martins, Adv. Synth. Catal. 2018, 360, 346.
- [7] D. R. White, J. T. Hutt, J. P. Wolfe, J. Am. Chem. Soc. 2015, 137, 11246.
- [8] D. Wang, L. Wu, F. Wang, X. Wan, P. Chen, Z. Lin, G. Liu, J. Am. Chem. Soc. 2017, 139, 6811.
- [9] T. H. Graham, C. M. Jones, N. T. Jui, D. W. C. MacMillan, J. Am. Chem. Soc. 2008, 130, 16494.
- [10] a) S. Yu, S. Ma, Angew. Chem. Int. Ed. 2012, 51, 3074; Angew. Chem. 2012, 124, 3128; b) B. Alcaide, P. Almendros, Chem. Soc. Rev. 2014, 43, 2886; c) J. Ye, S. Ma, Acc. Chem. Res. 2014, 47, 989; d) R. Santhoshkumar, C.-H. Cheng, Asian J. Org. Chem. 2018, 7, 1151.
- [11] For mechanistic studies on syn versus anti oxy- and azapalladation, see: a) R. M. Trend, Y. K. Ramtohul, E. M. Ferreira, B. M. Stoltz, Angew. Chem. Int. Ed. 2003, 42, 2892; Angew. Chem. 2003, 115, 2998; b) T. Hayashi, K. Yamasaki, M. Mimura, Y. Uozumi, J. Am. Chem. Soc. 2004, 126, 3036; c) R. M. Trend, Y. K. Ramtohul, B. M. Stoltz, J. Am. Chem. Soc. 2005, 127, 17778; d) A. B. Weinstein, S. S. Stahl, Angew. Chem. Int. Ed. 2012, 51, 11505; Angew. Chem. 2012, 124, 11673; e) N. J. Race, C. S. Schwalm, T. Nakamuro, M. S. Sigman, J. Am. Chem. Soc. 2016, 138, 15881; f) X. Kou, Q. Shao, C. Ye, G. Yang, W. Zhang, J. Am. Chem. Soc. 2018, 140, 7587.

[12] Z. Jiao, Q. Shi, J. S. Zhou, Angew. Chem. Int. Ed. 2017, 56, 14567; Angew. Chem. 2017, 129, 14759.

Angewandte

I Edition Chemie

- [13] For examples, see: a) Y. Zhang, M. S. Sigman, J. Am. Chem. Soc.
  2007, 129, 3076; b) K. H. Jensen, J. D. Webb, M. S. Sigman, J. Am. Chem. Soc. 2010, 132, 17471; c) X. Qi, C. Chen, C. Hou, L. Fu, P. Chen, G. Liu, J. Am. Chem. Soc. 2018, 140, 7415; d) C. Chen, P. M. Pflüger, P. Chen, G. Liu, Angew. Chem. Int. Ed. 2019, 58, 2392; Angew. Chem. 2019, 131, 2414.
- [14] N. G. Andersen, B. A. Keay, Chem. Rev. 2001, 101, 997.
- [15] For values of bite angles at palladium, see: a) C. Bolm, D. Kaufmann, S. Gessler, K. Harms, J. Organomet. Chem. 1995, 502, 47; b) A. H. Roy, J. F. Hartwig, Organometallics 2004, 23, 194.
- [16] J. Hu, Y. Lu, Y. Li, J. Zhou, Chem. Commun. 2013, 49, 9425.
- [17] P. Sarnpitak, K. Trongchit, Y. Kostenko, S. Sathalalai, M. P. Gleeson, S. Ruchirawat, P. Ploypradith, J. Org. Chem. 2013, 78, 8281.
- [18] C. Grandclaudon, V. Michelet, P. Y. Toullec, *Org. Lett.* **2016**, *18*, 676.
- [19] CCDC 1888827 (7 f), 1937818 (7g), 1937817 (7h), and 1880117 (7j) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre.
- [20] R. Willand-Charnley, T. J. Fisher, B. M. Johnson, P. H. Dussault, Org. Lett. 2012, 14, 2242.
- [21] A. Volkov, K. P. J. Gustafson, C.-W. Tai, O. Verho, J.-E. Bäckvall,
   H. Adolfsson, Angew. Chem. Int. Ed. 2015, 54, 5122; Angew.
   Chem. 2015, 127, 5211.
- [22] For examples, see: a) W. Yang, Y. Wang, J. Y. Roberge, Z. Ma, Y. Liu, R. M. Lawrence, D. P. Rotella, R. Seethala, J. H. M. Feyen, J. K. Dickson, *Bioorg. Med. Chem. Lett.* 2005, *15*, 1225; b) C. C. Musonda, G. A. Whitlock, M. J. Witty, R. Brun, M. Kaiser, *Bioorg. Med. Chem. Lett.* 2009, *19*, 401; c) J. Alen, M. Schade, M. Wagener, F. Christian, S. Nordhoff, B. Merla, T. R. Dunkern, G. Bahrenberg, P. Ratcliffe, *J. Med. Chem.* 2019, *62*, 6391.
- [23] For examples, see: a) E. R. Welin, A. A. Warkentin, J. C. Conrad, D. W. C. MacMillan, *Angew. Chem. Int. Ed.* 2015, 54, 9668; *Angew. Chem.* 2015, 127, 9804; b) Y. Kuang, X. Wang, D. Anthony, T. Diao, *Chem. Commun.* 2018, 54, 2558.

Manuscript received: September 18, 2019 Accepted manuscript online: November 6, 2019 Version of record online: December 19, 2019