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Umpolung of donor-acceptor cyclopropanes *via* N-heterocyclic carbene organic catalysis†

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A carbene-catalyzed formal umpolung of donor–acceptor (D–A) cyclopropanes is disclosed. The cyclopropane moiety is connected to an acetyl aldehyde that can be activated by a carbene catalyst. The initially electrophilic carbon attached to the donor group of the D–A cyclopropane aldehyde is inverted to form a nucleophilic reaction center. A subsequent reaction with isatins *via* a formal [3 + 2] process forms lactones bearing multiple functional groups with excellent enantio- and diastereoselectivities.

Introduction

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Cyclopropanes are structural moieties with unique reactivities in organic chemistry. They are useful building blocks for syntheses,¹ and can be used to probe reaction mechanisms such as those involving radical clock experiments.² However, despite the considerable ring strains existing in their tortuous chemical bonds, the C-C bonds in cyclopropanes are rather kinetically inert.3 Therefore, activated cyclopropanes have been rationally designed as one class of starting materials for ringextension and ring-opening reactions. In this regard, donoracceptor (D-A) cyclopropanes have been extensively studied as versatile building blocks for over half a century.⁴ Traditionally, D-A cyclopropanes are activated by Lewis acid catalysts,⁵ transition metal catalysts⁶ or organic catalysts⁷ to produce zwitterionic intermediates with an electrophilic carbon attached to the donor groups and a nucleophilic carbon attached to the acceptor groups (Fig. 1, right part). D-A cyclopropanes can also be activated through radical processes and electrochemical approaches.8 The umpolung activation of D-A cyclopropanes inverting the initial polarities of these two carbons (Fig. 1, left part), on the other hand, is much less studied. Progress in this direction has been dominated by transition

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metal catalysis and only a very limited number of examples have been reported to date.⁹

Organic catalysts provide robust tools for asymmetric transformations¹⁰ and can be used in the activation of D-A cyclopropanes for various stereoselective cycloaddition reactions (Fig. 2).7 For example, Jørgensen and co-workers disclosed the enamine-activated [2 + 2] cycloaddition reaction of D-A cyclopropanes in 2015 for the enantioselective synthesis of cyclobutanes (Fig. 2a, right part).^{7c} This strategy was further developed by Vicario and co-workers in 2016 in a domino process to prepare chiral pyrroloquinoline derivatives.7d The stereoselective 1,3-dipolar cycloaddition reaction of D-A cyclopropanes has also been realized by Jørgensen's group in 2017, with a chiral tertiary amine/thiourea multifunctional molecule applied as the organic catalyst (Fig. 2a, left part).^{7f} 1-Cyclopropylcarbaldehydes bearing multiple electron-withdrawing groups have been used as substrates in N-heterocyclic carbene (NHC) organocatalytic reactions (Fig. 2b, left part).¹¹ Ring-opening reactions of 1-cyclopropylcarbaldehyde molecules have been pioneered by Bode and co-workers in 2006,^{11b} with the corresponding esters and amides being afforded in excellent yields. Both intra- and inter-molecular esterification reactions using 1-cyclopropylcarbaldehyde with NHC catalysis have been developed by You,^{11f} Wang,^{11d,e,g} Ye,^{11i,k} Chi,^{11h} and others in the last decade.



Fig. 1 Catalytic activation of donor-acceptor cyclopropanes.



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Notably, for 1-cyclopropylcarbaldehydes used in all previous NHC catalysis studies, the aldehyde moiety and cyclopropane ring are directly connected with each other. To the best of our knowledge, the carbene-catalyzed activation of D–A cyclopropanes with the carbaldehyde moiety not directly attached to the cyclopropyl rings (2-cyclopropylacetaldehydes) has not been reported (Fig. 2b, right part).

Herein, we disclose that 2-cyclopropylacetaldehyde bearing a D–A cyclopropane moiety can be efficiently activated by NHC organic catalysts and it reacts with electrophiles through asymmetric cycloaddition reactions (Fig. 2c). The β -carbon of 2-cyclopropylacetaldehyde **1** is activated as a nucleophile to react with the isatin substrate 2 through a formal [3 + 2] cycloaddition reaction and gives the lactone product 3 in good to excellent yields and stereoselectivities. In this scenario, the initially electrophilic carbon attached to the donor group of the D–A cyclopropane (*e.g.*, in substrate **1** and the Breslow intermediate **I**¹²) is formally inverted to a nucleophilic carbon through the formation of acylazolium.

Results and discussion

We started the proposed umpolung activation of the D–A cyclopropanes by using **1a** and **2a** as the model substrates (Table 1). Several NHC catalysts were found to be efficient for this process, with the desired lactone products being afforded in moderate yields and with promising stereoselectivities (Table 1, entries 1 to 6). In particular, the aminoindanolderived NHC catalyst **F** bearing an *N*-Ph group with a NO₂ group installed on its indane unit¹³ could mediate the reaction

 Table 1
 Optimization of the reaction conditions^a



^{*a*} General conditions (unless otherwise specified): **1a** (0.15 mmol), **2a** (0.10 mmol), NHC (0.02 mmol), base (0.02 mmol), solvent (2.0 mL), 35 °C, 24 h. ^{*b*} Isolated yield of **3a**. ^{*c*} er was determined *via* HPLC on the chiral stationary phase. ^{*d*} dr was determined *via* ¹H NMR on the crude product. ^{*e*} **1a** (0.20 mmol), **2a** (0.10 mmol), F (0.02 mmol), Et₃N (0.02 mmol), CHCl₃ (0.5 mL), 35 °C, 24 h. Trt = triphenylmethyl. DMAP = 4-dimethylaminopyridine. DABCO = triethylenediamine.

to give product **3a** with excellent enantio- and diastereoselectivities (entry 6). Typical organic bases other than Et_3N tested here were inefficient for this reaction (*e.g.*, entries 7 to 8). The use of inorganic bases led to **3a** with low yields and slightly higher er values (*e.g.*, entries 9 to 10). Solvents such as CH_3CN , toluene, and CH_2Cl_2 did not perform well (*e.g.*, entries 11 to 13). Finally, the product yield could be improved to 82% when an excess amount of **1a** was used, with **3a** being formed with over 20:1 dr and 98:2 er (entry 14).

With acceptable conditions in hand, we next examined the reaction scope by using **1a** as the model D–A cyclopropane substrate to react with isatins bearing various substituents or substitution patterns (Scheme 1). Both electron-donating and electron-withdrawing groups could be installed at the 4- and 5-positions of the isatin substrates (2), with the corresponding lactone products **3** being afforded in moderate to good yields and with excellent enantioselectivities (**3b** to **3i**). Electron-withdrawing substituents were well tolerated at the 6-position of the isatins (**3j** to **3l**), while electron-donating substituents at the same position resulted in a decrease of the product yields (*e.g.*, **3m**). Both 4,6- and 5,6-difluoro-substituted isatins gave lactone products with excellent enantio- and diastereo-selectivities (**3n** to **30**). Replacing the *N*-Trt group with an *N*-Bn

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Scheme 1 Scope of isatins. Reaction conditions as stated in Table 1, entry 14. Yields are isolated yields after purification *via* SiO_2 column chromatography. er values were determined *via* HPLC on the chiral stationary phase. dr values were determined *via* ¹H NMR on the crude product.

group led to slight reduction of reaction yield and enantio-selectivity $(\mathbf{3p})$.

D-A cyclopropanes (1) bearing different acceptor groups were also tested (Scheme 2). The ethyl ester group in 1a could be switched to methyl and isopropyl ester groups without reduction in the product er values, although the yields slightly dropped (3q to 3r). Benzyl esters could also be installed on the D-A cyclopropane substrate to replace the ethyl esters of 1a, with the optically enriched product 3s being afforded with a moderate yield and diastereoselectivity. Replacing the di-ester group of 1a with a mono-electron-withdrawing cyanide group resulted in no desired reactions.

To gain insights into the reaction mechanism, several control experiments were performed (Fig. 3). 2-Cyclopropylcarbaldehyde **1a** could isomerize into the enal **1a1** under the catalytic conditions at the beginning of the cycloaddition reaction (Fig. 3a). Therefore, our catalytic umpo-



Scheme 2 Scope of D–A cyclopropanes. Reaction conditions as stated in Table 1, entry 14. Yields are isolated yields after purification via SiO_2 column chromatography. er values were determined via HPLC on the chiral stationary phase. dr values were determined via ¹H NMR on the crude product.

a) ¹H NMR spectra of control experiment:



b) proposed mechanism:



Fig. 3 $\,^{1}\mathrm{H}$ NMR studies on the reaction system and proposed reaction mechanism.

lung activation of 2-cyclopropylcarbaldehyde **1a** probably goes through a ring-opening/homoenolate activation cascade reaction pathway (Fig. 3b, Pathway A). However, the direct activation of the 2-cyclopropylcarbaldehyde **1a** by the NHC catalyst to provide the nucleophilic β -carbon cannot be ruled out at this stage (Fig. 3b, Pathway B).

In addition, the catalytic reaction of pre-prepared enal **1a1**¹⁴ and **2a** gave a much lower yield than reactions using 2-cyclo-



Fig. 4 Control experiments with different carbaldehydes.

propylcarbaldehyde **1a** as the substrate (Fig. 4a). Therefore, the use of the D-A cyclopropane aldehydes as the reaction substrates is practically favored for the preparation of chiral spirocyclic lactone products **3**. It is also worth mentioning that the umpolung activation of the D-A cyclopropane **1a2** was not successful under the current reaction conditions (Fig. 4b). The target lactone product **3t** was not observed, with both D-A cyclopropane **1a2** and isatin substrate **2a** being recovered in moderate to good yields. As a technical note, the D-A cyclopropane **1a2** could lead to complex side reactions and give unisolable decomposition/polymerization products in low yields.

The chiral lactone products obtained from our method possess several functional groups and are amenable for further transformations (Fig. 5). For example, the Trt protecting group on the nitrogen atom could be efficiently removed under acidic conditions to give unprotected lactam 5 in 70%



Fig. 5 Synthetic transformations.

yield and 95:5 er.¹⁵ The free NH group on 5 could be protected by a Boc group to give product 6 with good yield and enantioselectivity.¹⁶ The lactam 5 could also be acylated to give 7 in 94% yield with nearly no reduction in the product dr and er values.¹⁷ The lactone ring of the chiral product 3a could be opened by BnNH₂ *via* an ester-to-amide exchange to give 8 with a good yield and er value,¹⁸ although the diastereoselectivity slightly dropped. The acidic carbon of the malonate moiety on 3a could react with alkyl bromides¹⁹ (products 9 and 10) under basic conditions.

Conclusions

In summary, we have developed a formal umpolung activation of D–A cyclopropanes bearing aldehyde moieties *via* NHC catalysis. The initially electrophilic carbon attached to the donor group of the D–A cyclopropane is activated as the nucleophilic species to react with isatins. Lactone products bearing various substituents are afforded in moderate to good yields with generally excellent enantio- and diastereoselectivities. Further investigations into D–A cyclopropane activation reactions, especially those with applications in quick construction of sophisticated functional molecules, are currently in progress in our laboratories.

Conflicts of interest

There are no conflicts to declare.

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